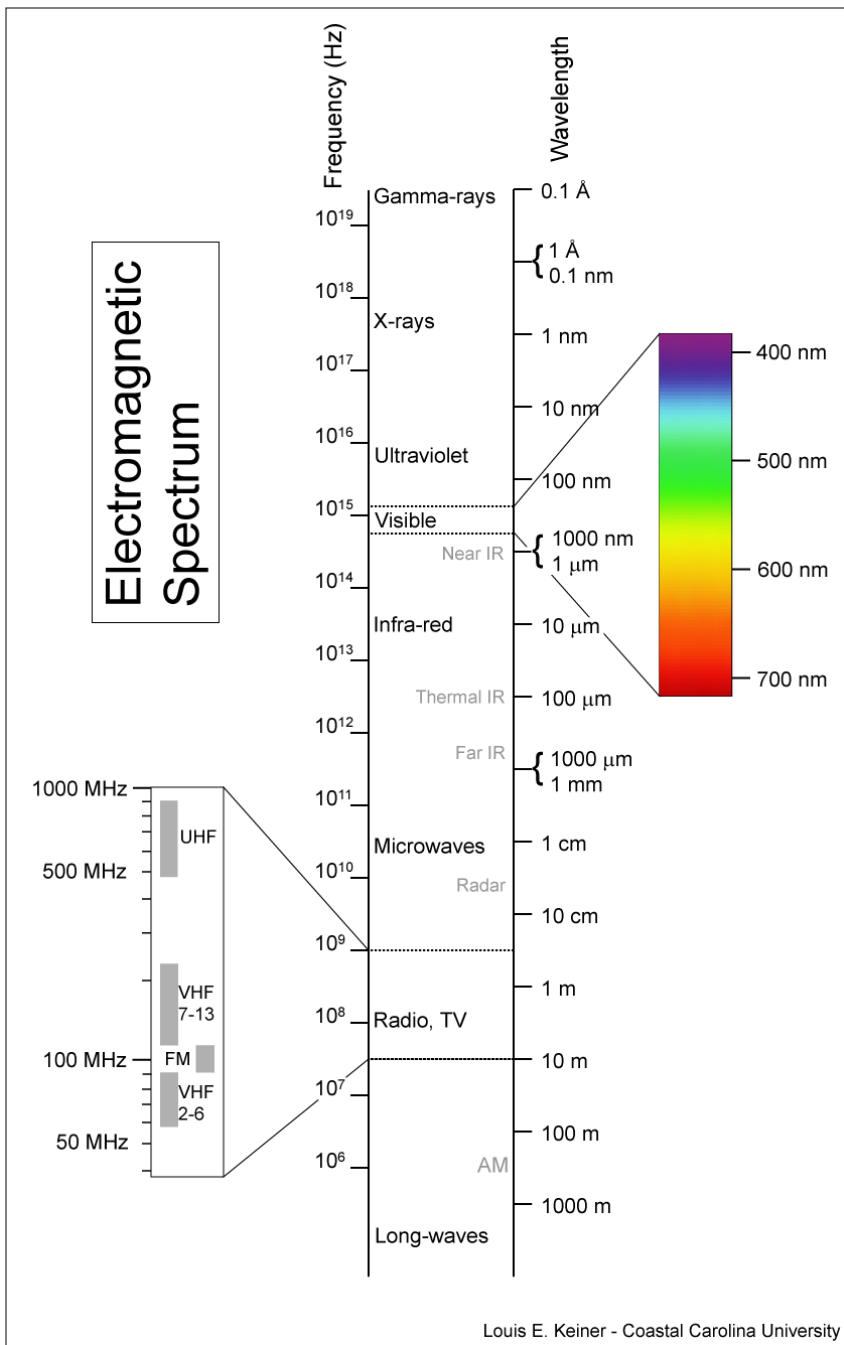


Introduction to  
Electromagnetic Radiation  
and  
Radiative Transfer

# Temperature Dice Results



Visible light, infrared (IR), ultraviolet (UV), X-rays,  $\gamma$ -rays, microwaves, and radio are all forms of electromagnetic radiation.

Each type has a different wavelength ( $\lambda$ ).

The shorter the wavelength, the greater the energy of individual photons (discrete packets of electromagnetic radiation).

Shortest to longest wavelength:  $\gamma$ -rays, X-rays, UV, visible, IR, microwave, radio

When a photon hits a molecule (gas, liquid, or solid), one of the following events will occur:

- The photon continues in the same direction (transmission)
- The photon goes in a new direction (scattering/reflection)
- The photon is destroyed, and its energy goes to the molecule (absorption)

The individual probabilities for transmission, scattering, or absorption depend on the molecule and the wavelength of the photon

A UV photon hitting a nitrogen molecule will probably be...

A UV photon hitting an ozone molecule will probably be...

A visible photon hitting a nitrogen molecule will probably be...

A visible photon hitting a water molecule will probably be...

An IR photon hitting a nitrogen molecule will probably be...

An IR photon hitting a water molecule will probably be...

- transmitted?
- scattered?
- absorbed?

## Molecules can also emit photons

- Molecules that are good absorbers at a particular wavelength are also good emitters (potentially) at that wavelength
- The probability of emission also depends on the temperature of the molecules

## Characteristics of a blackbody:

- A blackbody is a substance with 100% absorption at every wavelength (no transmission or scattering)
- A blackbody emits the maximum amount of radiation possible
- Warm blackbodies emit more radiation than cold blackbodies
- The wavelength with the most emission is shorter for warmer blackbodies

$$F = \sigma T^4$$

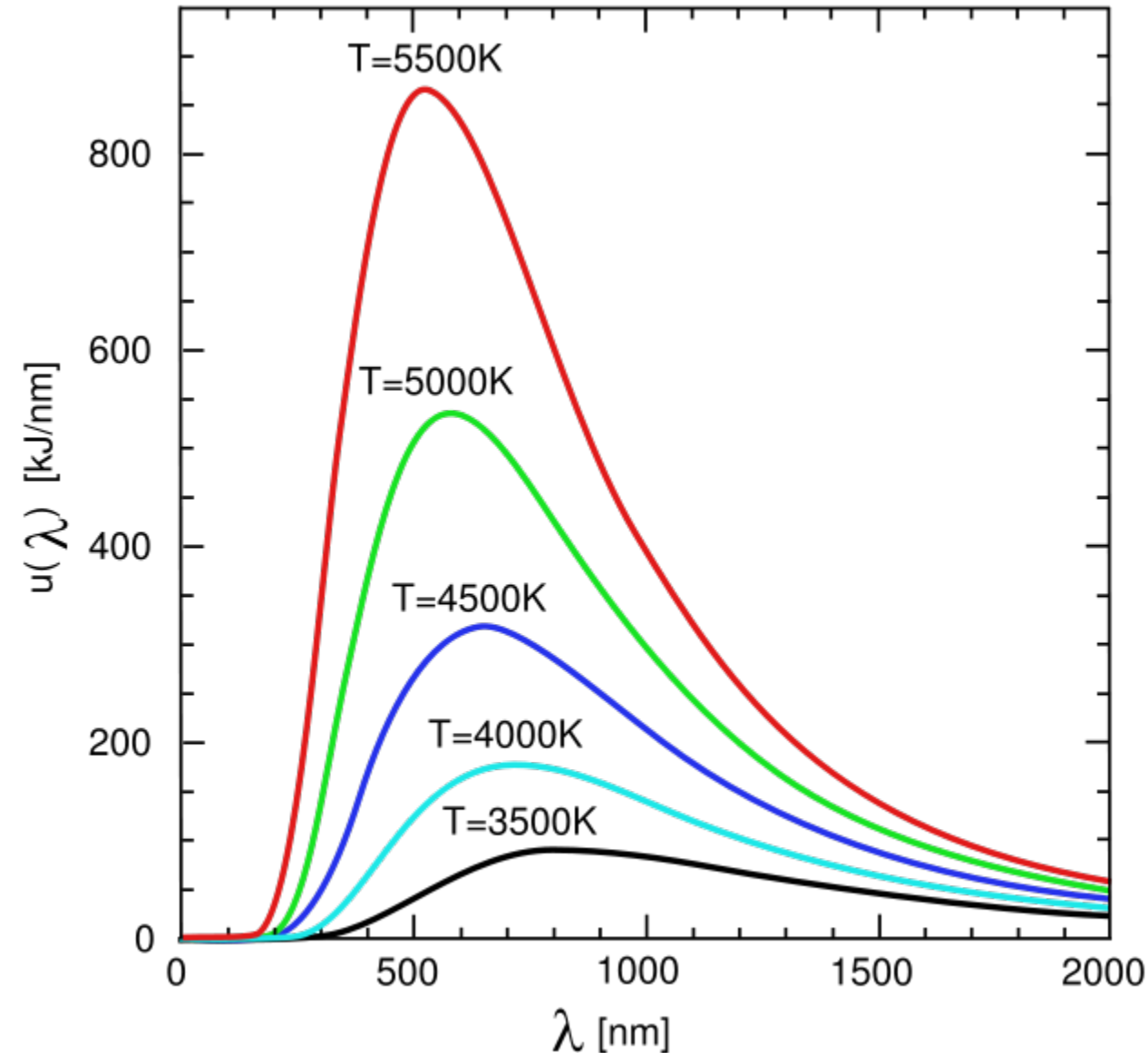
$F$  = radiation energy flux summed over all wavelengths (units are  $\text{W m}^{-2}$ )

$T$  = temperature (units are K)

$$\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$$

$$\lambda_{max} = b / T$$

$$b = 2898 \text{ } \mu\text{m K}$$



Author: 4C



Approximate the Sun as a blackbody:

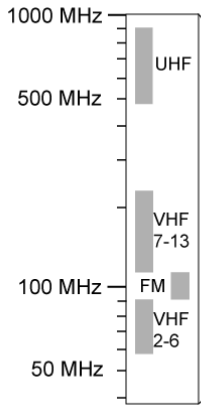
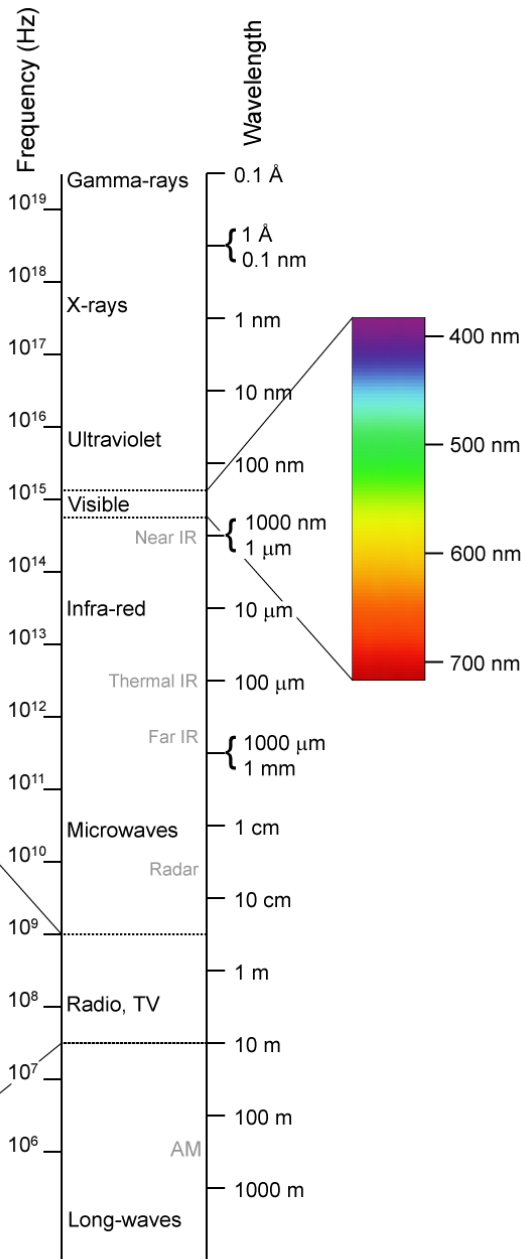
- Temperature at surface is about 5800 K
- Using  $F = \sigma T^4$ , flux at surface is about  $6.4 \times 10^7 \text{ W m}^{-2}$   
*(This is equivalent to more than 1 million 60W light bulbs every square meter)*
- Using  $\lambda_{max} = b / T$ , what is the wavelength of most emission?

*(b is about 3000  $\mu\text{m K}$ )*

Approximate the Sun as a blackbody:

- Temperature at surface is about 5800 K
- Using  $F = \sigma T^4$ , flux at surface is about  $6.4 \times 10^7 \text{ W m}^{-2}$   
*(This is equivalent to more than 1 million 60W light bulbs every square meter)*
- Using  $\lambda_{max} = b / T$ , wavelength of most emission is about  $0.50 \text{ } \mu\text{m}$   
*(This is in the middle of the visible spectrum)*

# Electromagnetic Spectrum



$$1000 \text{ nm} = 1 \mu\text{m}$$

Approximate the Earth as a blackbody:

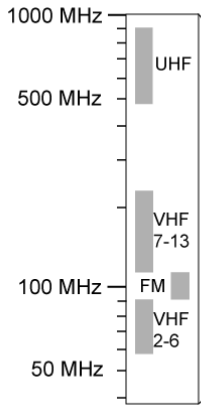
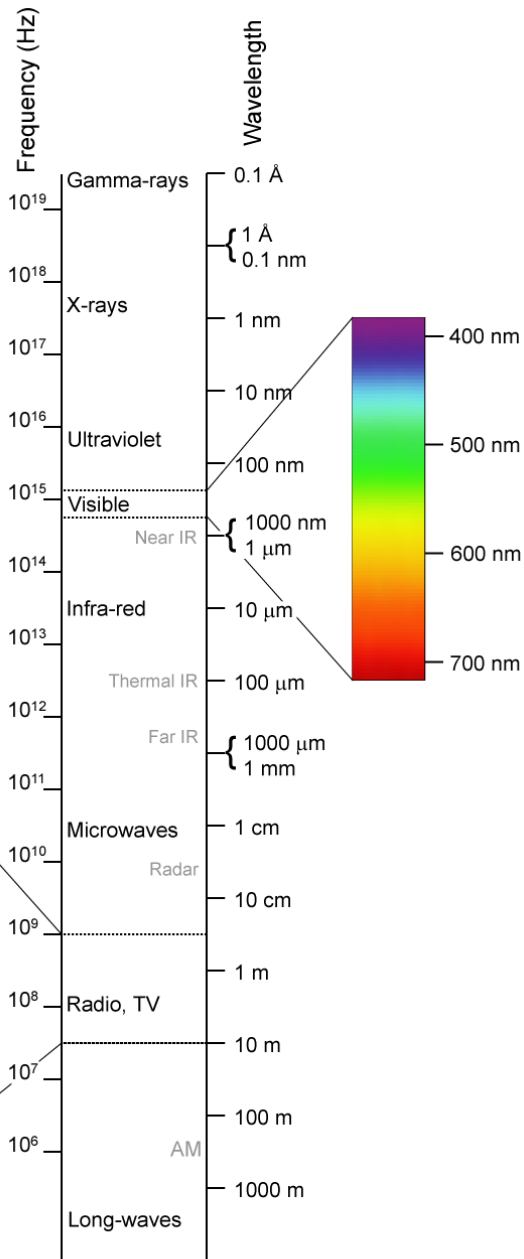
- Temperature at surface is about 288 K
- Using  $F = \sigma T^4$ , flux at surface is about  $390 \text{ W m}^{-2}$   
*(This is equivalent to more than four and a half 60W light bulbs every square meter)*
- Using  $\lambda_{max} = b / T$ , what is the wavelength of most emission?

*(b is about  $3000 \mu\text{m K}$ )*

Approximate the Earth as a blackbody:

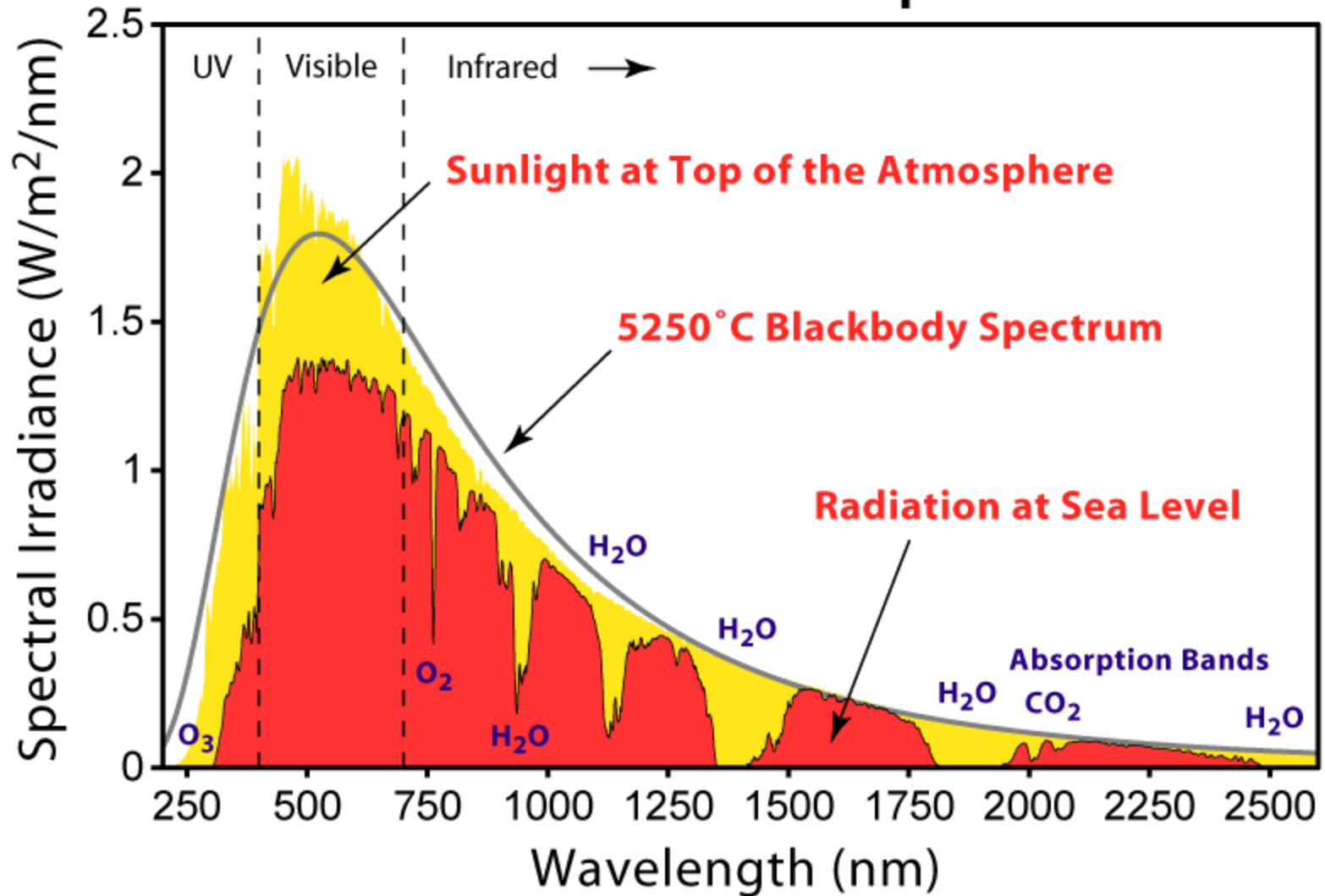
- Temperature at surface is about 288 K
- Using  $F = \sigma T^4$ , flux at surface is about  $390 \text{ W m}^{-2}$   
*(This is equivalent to more than four and a half 60W light bulbs every square meter)*
- Using  $\lambda_{max} = b / T$ , wavelength of most emission is about  $10 \text{ }\mu\text{m}$   
*(This is in the middle of the thermal IR spectrum)*

# Electromagnetic Spectrum



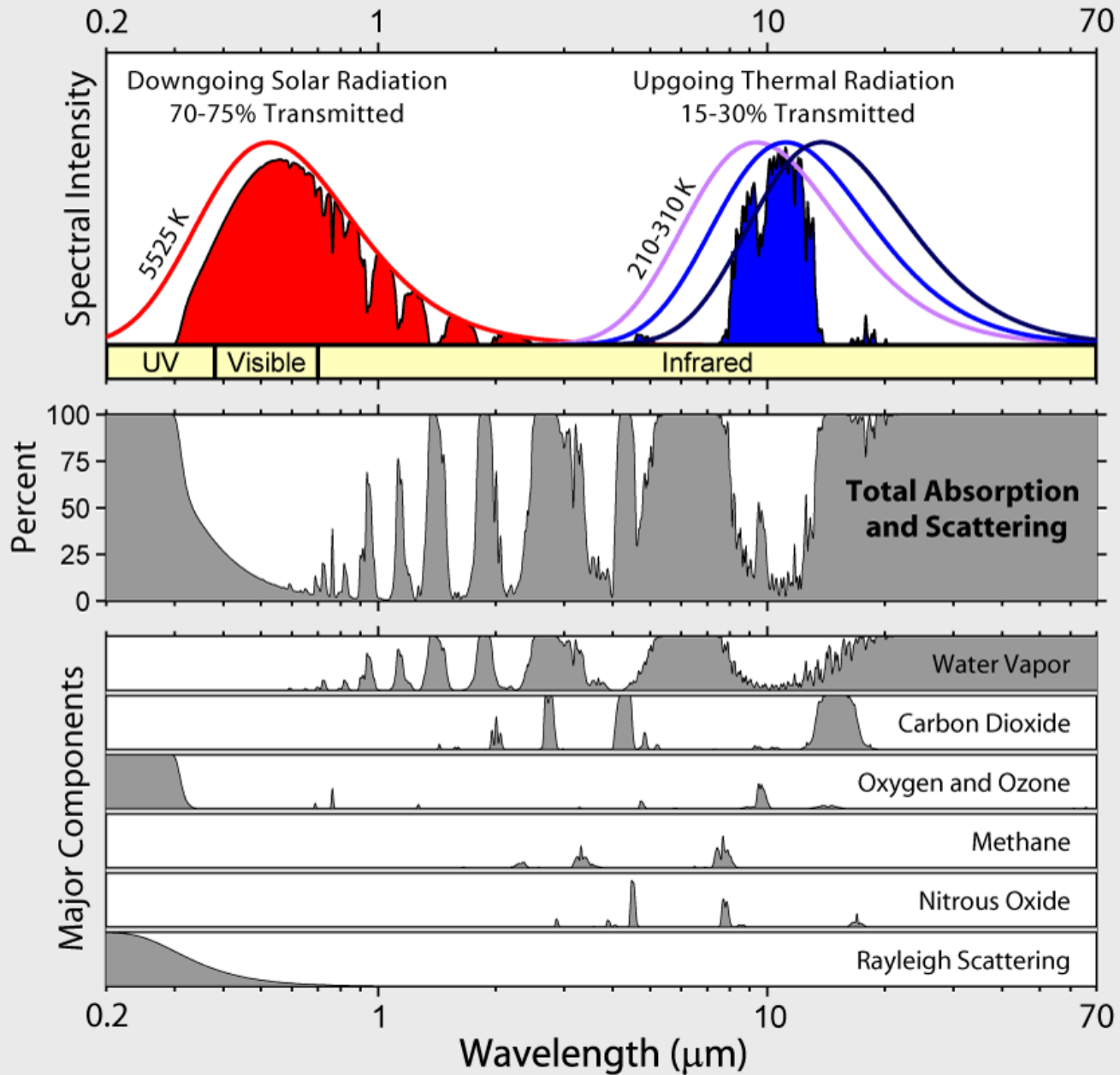
$$1000 \text{ nm} = 1 \mu\text{m}$$

# Solar Radiation Spectrum



From <http://www.globalwarmingart.com>

# Radiation Transmitted by the Atmosphere



From <http://www.globalwarmingart.com>



The individual probabilities for transmission, scattering, or absorption depend on the molecule and the wavelength of the photon

A UV photon hitting a nitrogen molecule will probably be...

A UV photon hitting an ozone molecule will probably be...

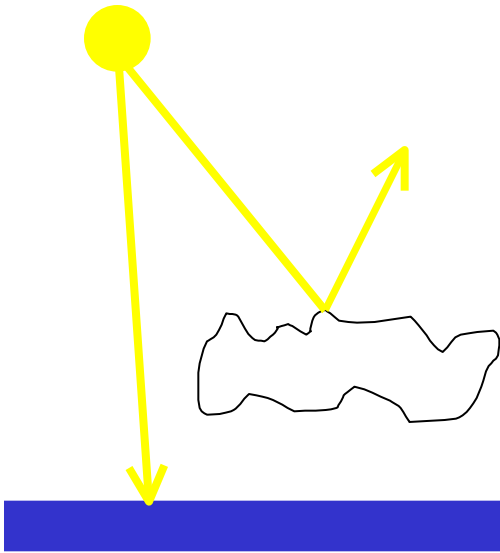
A visible photon hitting a nitrogen molecule will probably be...

A visible photon hitting a water molecule will probably be...

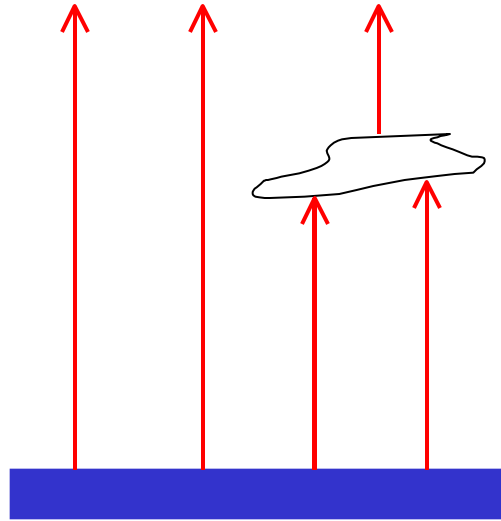
An IR photon hitting a nitrogen molecule will probably be...

An IR photon hitting a water molecule will probably be...

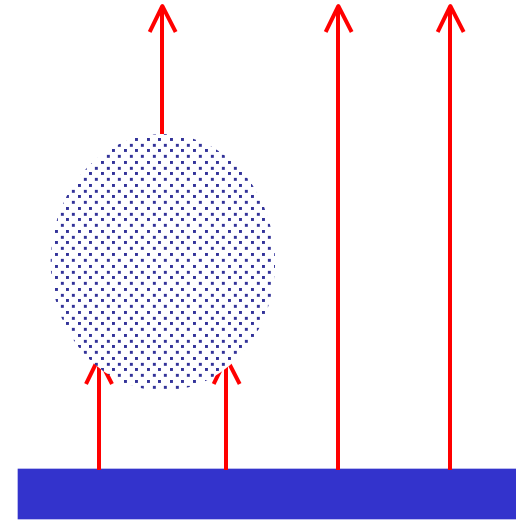
- transmitted?
- scattered?
- absorbed?



visible



infrared (water  
vapor window)



infrared (water  
vapor)

# Satellite Images

- Visible (VIS): white indicates large radiation flux, dark indicates small radiation flux
- IR water vapor window (IR): white indicates small radiation flux, dark indicates large radiation flux
- IR Water vapor (WV): white indicates small radiation flux, dark indicates large radiation flux