

Aerosol Dynamics Model

Four Lognormal Modes

- Lognormal distribution and properties
 - e.g. Volume-mean for lognormal distribution
- Four lognormal mode algorithm
 - Coarse mode
 - Accumulation mode
 - Condensation mode
 - Nucleation mode

MATLAB Lognormal Distribution

```

for dp=1:NUMBERsections %loop to assign data points to lognormal fit
Dp(dp)=MINdiameter*(MAXdiameter/MINdiameter)^(dp/NUMBERsections);
%increments the diameter over the plotted range for the specified number of sections
%Equation for lognormal distribution, [Eqn. 7.33, Seinfeld and Pandis, 1998]
dNdlndp(dp)=(N/(((2*pi)^0.5)*log(seg))) *exp((-log(Dp(dp))-log(Dpg))^2)/(2*log(seg)^2));
end
    
```

- A loop is used to assign each element of the vectors Dp and dNdlndp to a value

MATLAB Lognormal Mean Volume Diameter

```

for dp=1:NUMBERsections %loop to assign data points to lognormal fit
Dp(dp)=MINdiameter*(MAXdiameter/MINdiameter)^(dp/NUMBERsections);
%increments the diameter over the plotted range for the specified number of sections
%Equation for lognormal distribution, [Eqn. 7.33, Seinfeld and Pandis, 1998]
dNdlndp(dp)=(N/(((2*pi)^0.5)*log(seg))) *exp((-log(Dp(dp))-log(Dpg))^2)/(2*log(seg)^2);
%Equation for volume-mean diameter of lognormal distribution, [Eqn. 7.52, Seinfeld and Pandis, 1998]
DpgV=exp(log(Dpg)+3*(log(seg))^2);
end
    
```

- Mean volume diameter is needed to get mass
- Mean volume diameter is larger than D_{pg}

MATLAB Critical Cluster Size: Four Lognormal-Mode Approach

```

%Run Conditions -- these vary depending on the time and place you are modeling
NCoars=100; %total particle number concentration of accumulation mode in #/cm^3
DpgCoars=5; %geometric mean particle diameter of accumulation mode in micron
sgCoars=1.5; %"sigma-g" or width of accumulation mode distribution
NAccum=100; %total particle number concentration of accumulation mode in #/cm^3
DpgAccum=0.1; %geometric mean particle diameter of accumulation mode in micron
sgAccum=1.5; %"sigma-g" or width of accumulation mode distribution
NCond=100; %total particle number concentration of condensation mode in #/cm^3
DpgCond=0.1; %geometric mean particle diameter of condensation mode in micron
sgCond=1.5; %"sigma-g" or width of condensation mode distribution
NNucle=1; %total particle number concentration of nucleation mode in #/cm^3
DpgNucle=0.005; %estimated geometric mean particle diameter of nucleation mode including H2SO4, H2O in micron
sgNucle=1.1; %estimated "sigma-g" or width of nucleation mode distribution
-
for dp=1:NUMBERsections %loop to assign data points to lognormal fit
%Equation for lognormal distribution, [Eqn. 7.33, Seinfeld and Pandis, 1998]
nNdlndp(dp)=...
+(NCoars/(((2*pi)^0.5)*log(segCoars))) *exp((-log(Dp(dp))-log(DpgCoars))^2)/(2*log(segCoars)^2)...
+(NAccum/(((2*pi)^0.5)*log(segAccum))) *exp((-log(Dp(dp))-log(DpgAccum))^2)/(2*log(segAccum)^2)...
+(NCond/(((2*pi)^0.5)*log(segCond))) *exp((-log(Dp(dp))-log(DpgCond))^2)/(2*log(segCond)^2)...
+(NNucle/(((2*pi)^0.5)*log(segNucle))) *exp((-log(Dp(dp))-log(DpgNucle))^2)/(2*log(segNucle)^2);
end
    
```

Deposition

- Deposition rate matlab code
- Settling velocity
- Deposition only size distribution evolution
 - Summary for 96 hr

MATLAB Deposition Rate

```

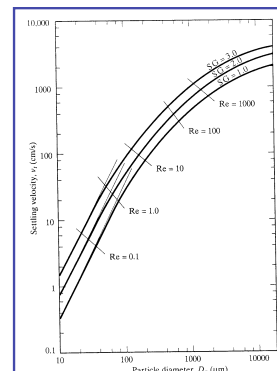
%Deposition Rate
Kn=2*AirMeanFreePath/(Dpg*CMperMIC); %Knudsen number [Seinfeld and Pandis, Eq. 8.1]
Cc=1+Kn*(1.257+0.4*exp(-1.1/Kn)); %Cunningham correction factor [Seinfeld, 1968, p. 317]
ParticleMass=(Dpg*CMperMIC)^3*pi/6*ParticleDensity; %mean particle mass in grams
Vt=(ParticleMass/GperKG)*GRAVITY/Cc/(3*pi*AirViscosity*Dpg*CMperMIC);
%terminal velocity [Seinfeld, 1968, p. 317]
DepnRate=(N*Vt/Height)*SSCperHR; %deposition rate in terms of particles/cm3/hr

%Calculate mass and number changes here
N=N-DepnRate*t;
%new number after loss of particles due to deposition, assuming 1 hr time step
    
```

- 1st calculate deposition rate
- 2nd subtract from previous number

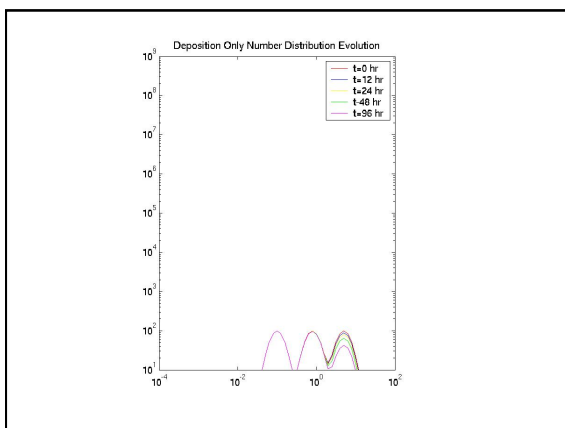
Settling Velocity

- terminal velocity of the particle in this fluid, v_t , where the particle has reached steady state



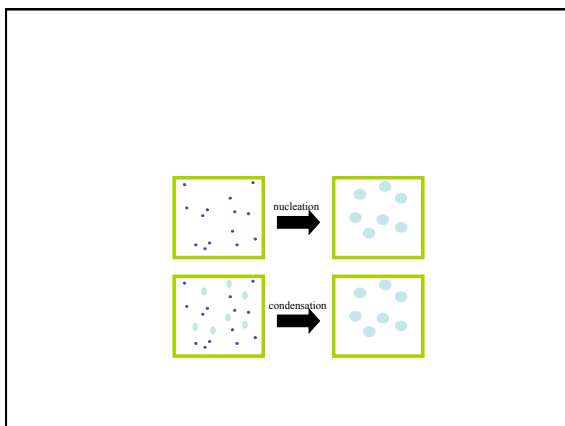
for unit density spheres in air at 20°C

D_p (μm)	τ (sec)	v_t (m sec ⁻¹)
0.1	9.2×10^{-8}	9.0×10^{-7}
1.0	3.6×10^{-6}	3.5×10^{-5}
10.0	3.1×10^{-4}	3.0×10^{-3}



Condensation

- Condensation rate matlab code
 - Vapor-phase loss
 - Particle-phase growth
- Condensation rates
- Condensation only size distribution evolution
 - Summary for 96 hr



MATLAB Particle and Vapor Changes

```

%Calculate mass and number changes here for each mode
DeltaN=(-DepnRateCoars)*tSTEP;
NCoars=NCoars+DeltaN %new number after deposition loss of particles
DeltaN=(-DepnRateAccum)*tSTEP;
NAccum=NAccum+DeltaN %new number after deposition loss of particles
DeltaN=(-DepnRateCond)*tSTEP;
NCond=NCond+DeltaN %new number after deposition loss of particles
DeltaN=(-DepnRateNucl+NuclRate)*tSTEP;
%Nucln=Nucln+DeltaN %new number after deposition loss and nucleation gain of particles
#2304(t+2)=#2304(t+1)-NuclRate*VaporPerParticle*tSTEP
    
```

- The sulfuric acid that nucleates new particles is removed from the vapor phase

Coagulation

- **Coagulation** is growth by collision of particles
 - decreases particle number
 - increases particle size
 - no net change in particle mass

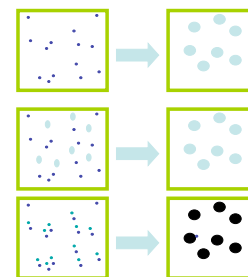


- the process of coagulation is almost entirely controlled by the physical properties of a particle, namely the particle ambient **diameter**

Nucleation

- Nucleation rate matlab code
- Nucleation rates
- Nucleation only size distribution evolution
 - Initial 0 hr and 12 hr
 - Summary for 96 hr

- Growth of clusters
- **homogeneous**=single phase (gas→particles)
- **heterogeneous**=two phases (gas+particles→big particles)
- **heteromolecular**=more than one species of vapors nucleate together (two kinds of gases→particles)

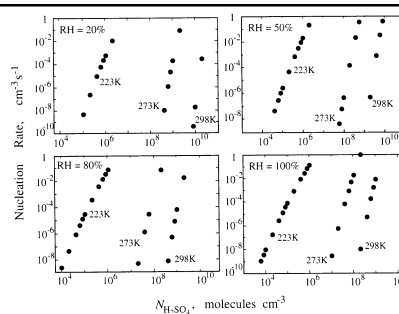


MATLAB Nucleation Rate

```

%Run Conditions -- these vary depending on the time and place you are modeling
H2SO4(1)=0.001*2.463e10; %initial mixing ratio of H2SO4 vapor in molec/cm3
VaporPerNuclParticle=(.005/.001)*3*100; %estimated number of molecules H2SO4 vapor in each particle
-
%Nucleation Rate for nucleation mode
NuclRate=(3.38e6+2.45e8*(H2SO4(t+1)^0.147))*SE0perHR
% nucleation rate at 273K and 80%RH in p/cm3/hr from Seinfeld and Pandis, Fig. 10.11
if ((NuclRate*VaporPerNuclParticle*tSTEP)>H2SO4(t+1))
% check if all vapor nucleates, determine max. possible rate
NuclRate=(H2SO4(t+1)/VaporPerNuclParticle)/tSTEP
end
    
```

- To evaluate nucleation, the model needs humidity (p_a) and acidity (N_a) plus temperature (T)



- Rate depends exponentially on temperature

$$J = (4\pi R_p^2) \frac{p_a}{(2\pi m_p kT)^{3/2}} N_a \exp\left(-\frac{\Delta G^*}{kT}\right)$$
 - Faster at colder temperatures

Kulmala and Laaksonen, 1990

